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# Preparation and Characterization of Polyurethane and its Composites based on Cardanol Modified with Corn Starch and Coir Fiber

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**Abstract:** This study presents the preparation of novel polyurethane and its composites from renewable resources of cardanol, corn starch and coir fiber. Corn starch was converted into glycoside through transglycosylation reaction with ethylene glycol. The prepared glycol-glycoside was analyzed by specific gravity, viscosity, hydroxyl value, molecular weight and FT-IR spectral study. Cardanol based-polyol was synthesized by condensing cardanol with formaldehyde in the mole ratio 1:5 using melodic acid as catalyst and subjected to epoxidation followed by hydrolysis. The glycol-glycoside were incorporated to the cardanol based-polyol with 4,4'-methylenebis(cyclohexyl) isocyanate using coir fiber under NCO/OH molar ratio constant at 1.4. The products were characterized by spectral properties and thermal properties. These results reveal the high crosslink performance of polyurethane and its composites due to the presence of glycol-glycoside.

**Keywords:** Corn starch; glycol-glycoside; cardanol; 4,4'-methylenebis(cyclohexyl) isocyanate; coir fiber; polyurethane composites.

## I. INTRODUCTION

The use of renewable raw materials in the polyurethane synthesis has been given significant recently and vast efforts have been made in the search of new economically sustainable and environmentally friendlier alternatives to petrochemicals which are gradually cost [1]. Starch is a natural polymer derived from corn and other crops possesses many unique properties and completely biodegradable in a wide variety of environments [2-6].

Transglycosylation reaction between starch and ethylene glycol produce a mixture of isomeric glycol glycosides that exhibits a potential use as a polyol raw material for surfactant and rigid urethane foam manufacture [7]. The starch-based polymers are potential for applications in biomedical and environmental fields [8,9]. Among renewable raw materials, cardanol is a phenolic compound with a C<sub>15</sub> unsaturated aliphatic side chain in the meta position [10] obtained from cashew nut shell liquid [11]. It has been widely used for making polyurethanes, polyesters, polyamides, composites, etc. [12-15].

Polyurethane is one of the polymer products made by reacting polyol with an isocyanate [16,17] showing better thermal stability and mechanical properties [18]. Polyurethane composites prepared from cardanol and formaldehyde in the presence of catalyst, in different molar ratios and natural fibers [19].

Coir fiber is the thickest and most resistant of all commercial natural fiber exhibit high tensile strength, high toughness, low density and recyclable [20,21]. Fiber-reinforced composites are finding applications in diverse fields from appliances to spacecrafts [22].

## II. EXPERIMENTAL

### A. Materials and methods

#### 1) Materials

Corn starch was purchased from Thermo Fisher Scientific India Pvt. Ltd., Mumbai. Ethylene glycol was obtained from Merck, Mumbai. Cardanol was obtained from M/s Satya Cashew Chemicals Pvt. Ltd., Chennai. Formaldehyde (40% solution), and malonic acid and methanol were received from Merck, Mumbai. 4,4'-methylenebis(cyclohexyl) isocyanate and dibutyltin dilaurate was received from Aldrich Chemicals, Bangalore. The chemicals were used as received. Coir fiber was procured from local sources.

#### 2) Alkali treatment of coir fiber

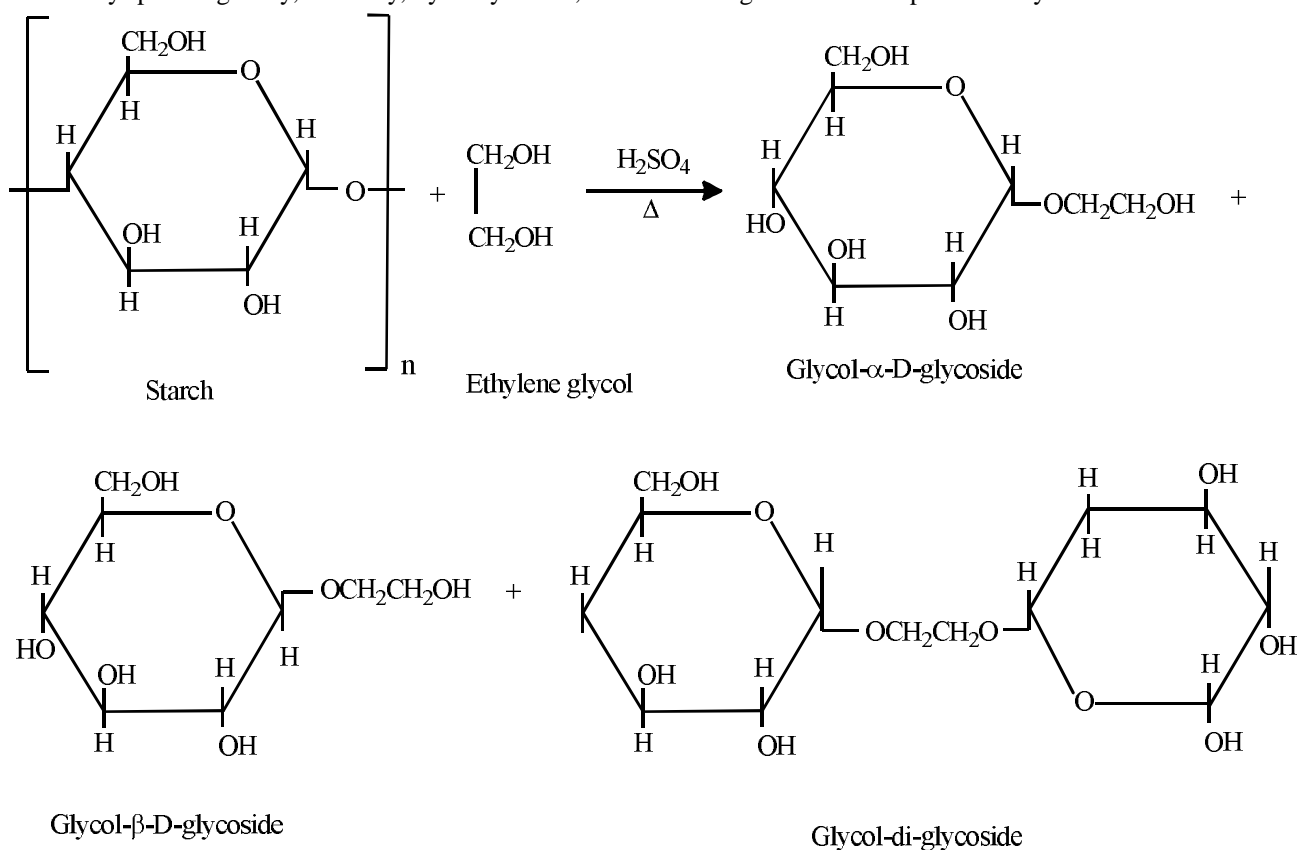
The coir fiber was cleaned initially and dried in atmosphere for one day and then kept in an oven at 50°C to remove the moisture. Then coir fiber was treated in 10% NaOH solution for 1 h followed by washing with distilled water and drying for 24 h at 60°C. The dried fiber was cut into 5 mm length prior to use [23,24].

### 3) Methods

Infrared spectra of the polyurethane and its composites were taken in a Shimadzu FT-IR-8400S spectrometer by KBr pellet method. Thermo gravimetric analysis (TGA)/differential thermal analysis (DTA) with DSC calculation, SIINTat a rate of 10 K/min in air/nitrogen were studied.

#### B. Synthesis of glycol-glycoside

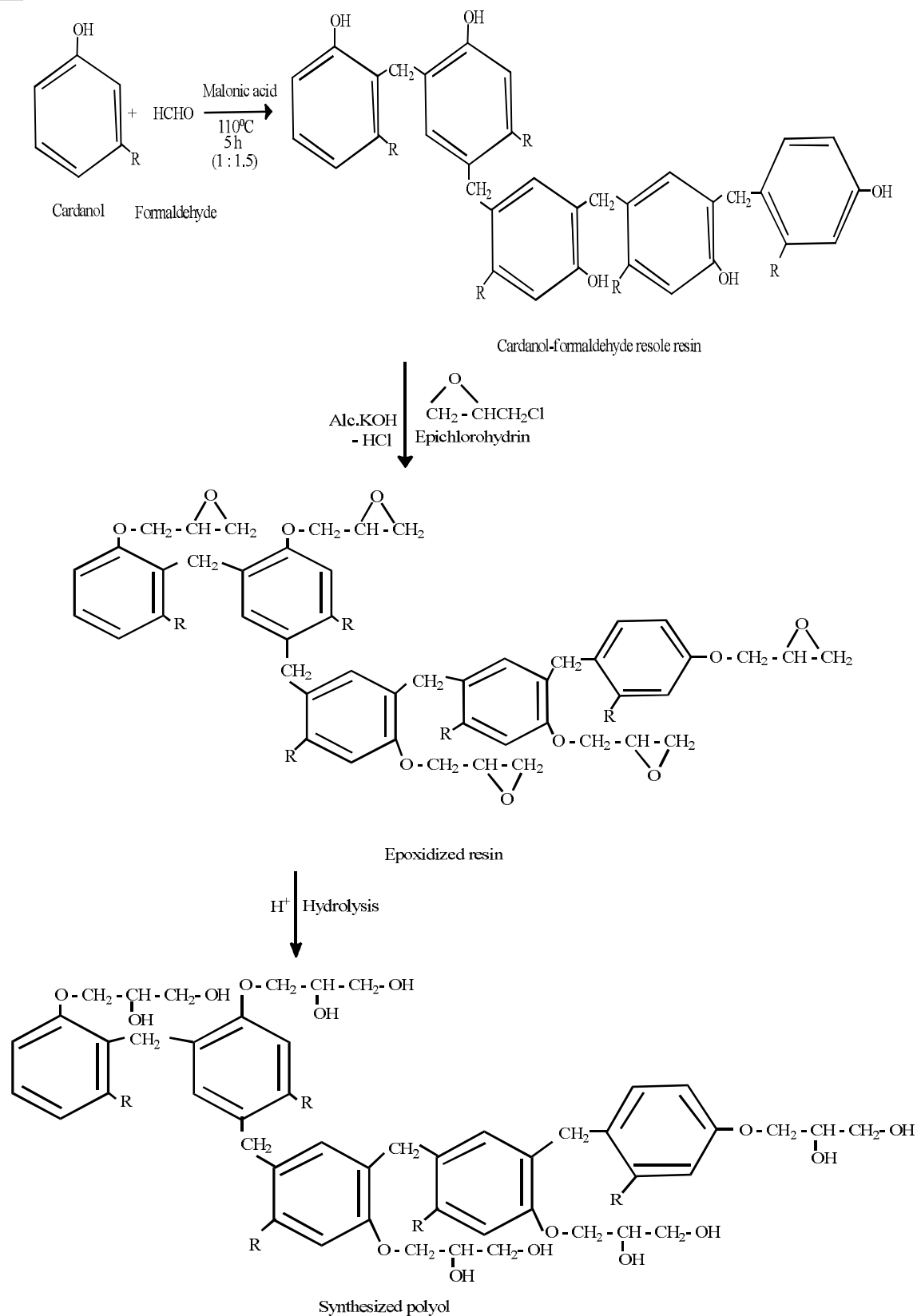
Ethylene glycol was taken in a three-necked flask equipped with a reflux condenser, thermometer and nitrogen blanket with good stirring and added sulphuric acid and heated to 120°C. Then corn starch(dry basis) was added over a 40-minute period while maintaining a temperature of 120°C. When addition was completed, reduce the pressure to 40 mm Hg and continue heating and stirring for 30 minutes. Then, the mixture was cooled and neutralized the sulphuric acid with calcium carbonate. Again heated and removed the un reacted ethylene glycol at 5 to 10 mm Hg as the temperature was slowly raised to about 140°C. After stripping was completed, dissolved the glycoside in water, decolorized with charcoal, filtered and finally, adjusted the concentration to 70-80% solids in water. The prepared glycol-glycoside was clear and have a colour of Gardner 1 [25]. The glycol-glycoside was characterized by specific gravity, viscosity, hydroxyl value, molecular weight and FT-IR spectral analysis.



Scheme 1. Reaction of glycosylation of starch using ethylene glycol

#### C. Synthesis of cardanol based-polyol

Cardanol was taken in a three necked flask equipped with a Liebig condenser, mechanical stirrer and thermometer. Formaldehyde and 1% malonic acid catalyst in methanol was added to the cardanol through a dropping funnel. The reaction was carried out at temperature  $110^\circ \pm 5^\circ\text{C}$  for 5 h. Then, the resin was precipitated in distilled water and purified by dissolving in ether. Major fractions are collected and dried using a rotary evaporator under vacuum. Epichlorohydrin was added to the synthesized resole resin. The reaction was carried out at temperature  $60^\circ\text{C}$  for 4 h. The obtained epoxidized resin are neutralized with potassium hydroxide, filtered and washed with distilled water several times. Then the resin was purified by dissolving in ether. After purifying, methanol was added to the epoxidized resin. The reaction was carried out at temperature  $60^\circ\text{C} \pm 5^\circ\text{C}$  for 30 min. The reaction mixture was checked using thin layer chromatographic method. Thus, the formed cardanol based-polyol (synthesize dpolyol) was analyzed.

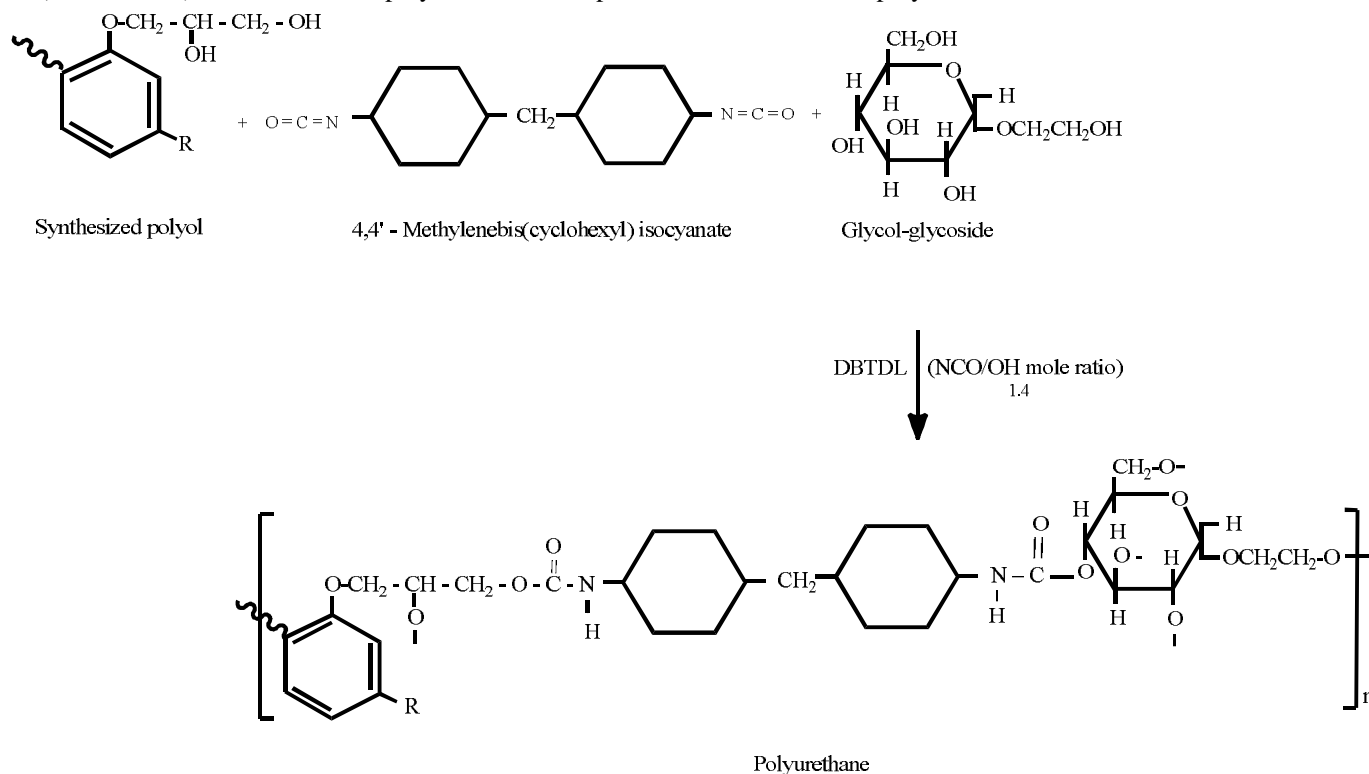


Where, R = C<sub>15</sub>H<sub>31</sub>-n

Scheme 2. Formation of cardanol based-polyol

**D. Synthesis of polyurethane and its composites based on cardanol based-polyol, glycol-glycoside using coir fiber**

Novel polyurethane (CDPGPU) was prepared from cardanol based-polyol using glycol-glycoside (10%), 4,4'-methylenebis(cyclohexyl) isocyanate and dibutyltin dilaurate (0.12 wt. %) as catalyst keeping the isocyanate index (NCO/OH mole ratio) constant at 1.4 (Scheme 3.5). The composites (CDPGPU 5 and CDPGPU 10) have been fabricated by incorporating the coir fiber (5% and 10%) into the obtained polyurethane. The spectral and thermal of the polyurethane sheets were evaluated.



Scheme 3. Formation of polyurethane based on glycol-glycoside

**III. RESULTS AND DISCUSSION**

**A. Characterization of glycol-glycoside**

1) *Physico-chemical properties of glycol-glycoside:* Glycol-glycoside possess pale yellow colour and odorless. The relatively higher specific gravity and viscosity are due to the content of glycoside. The determination of molecular weight, hydroxyl number and FT-IR spectral analysis leads to predict the structure of glycol-glycoside as given in Table 1.

Table 1. Physico-chemical properties of glycol-glycoside

Properties	Glycol-glycoside
Colour	Pale yellow
Odour	Odourless
Specific gravity(g/cc at 30°C)	0.8519
Viscosity at 30°C (cps)	153
Hydroxyl value	182
Number of hydroxyl groups	5
Molecular weight	264

2) FT-IR spectroscopy of glycol-glycoside

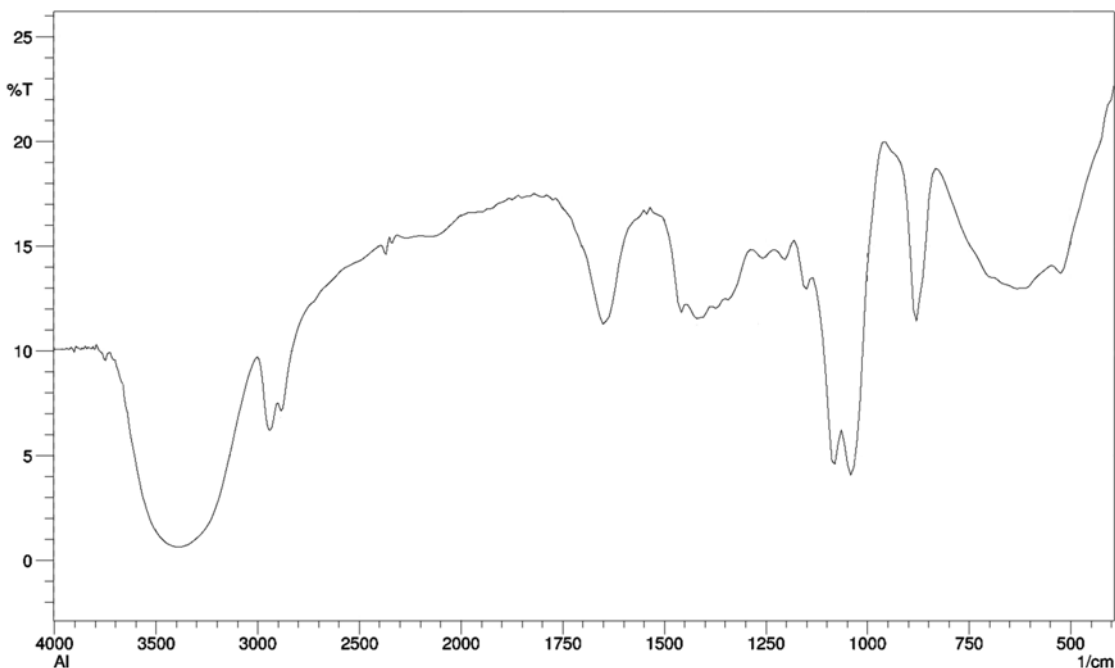


Figure 1. FT-IR spectrum of glycol-glycoside

FT-IR spectrum (Figure 1) of the synthesized glycol-glycoside show the sharp bands at 1080 and 1041  $\text{cm}^{-1}$  are due to -O-C-O- linkage confirms glycosylation of starch. -OH stretching band at 3387  $\text{cm}^{-1}$  reveals the presence of free hydroxyl groups in it. The bands at 2939 and 2885  $\text{cm}^{-1}$  are due to -CH stretching.

B. Characterization of polyurethane and its composites

1) FT-IR spectroscopy of polyurethane and its composites: The FT-IR spectra of the polyurethane and its composites (Figure 2 – Figure 4) showed characteristic absorption at 3305-3315  $\text{cm}^{-1}$  corresponding to urethane linkage (-NH stretching, bonded). The peaks at 1720-1745  $\text{cm}^{-1}$  indicates C=O stretching (free) in urethane, 1635-1650  $\text{cm}^{-1}$  corresponding to C=O stretching (bonded) in urethane and 1530-1550  $\text{cm}^{-1}$  indicates N-H bending in urethane. The peak at 1020-1110  $\text{cm}^{-1}$  is due absorption band of the -O-C-O-group developed by the glycosylation of starch.

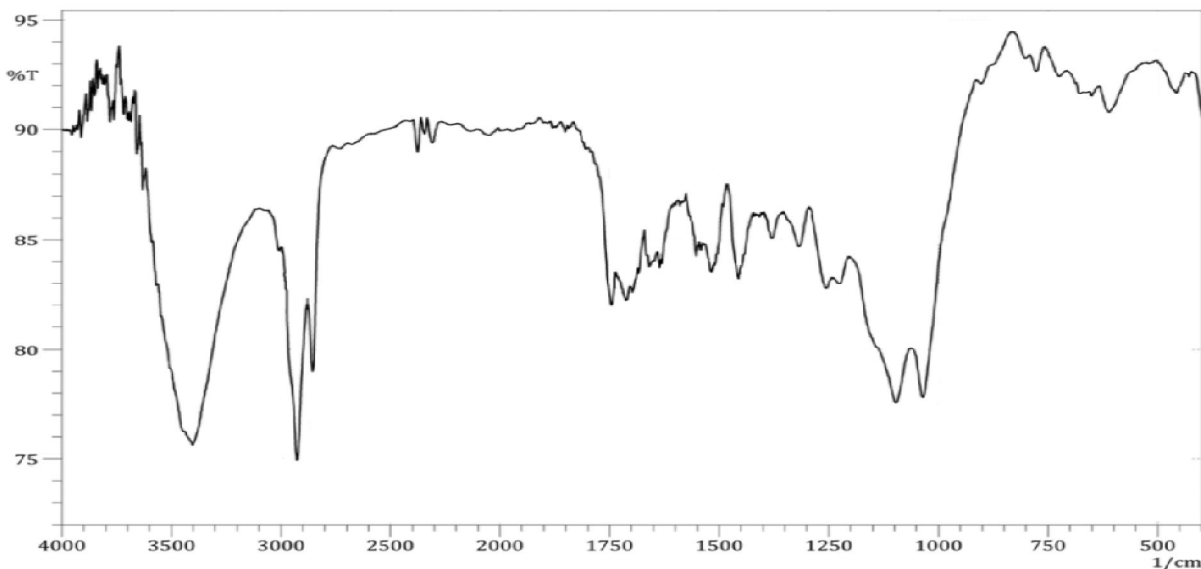


Figure 2. FT-IR spectrum of CDPGPU

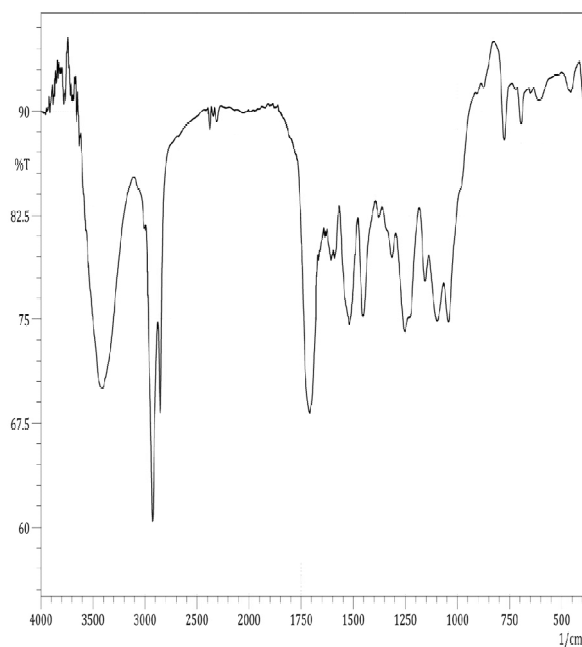


Figure 3.FT-IR spectrum of CDPGPU 5

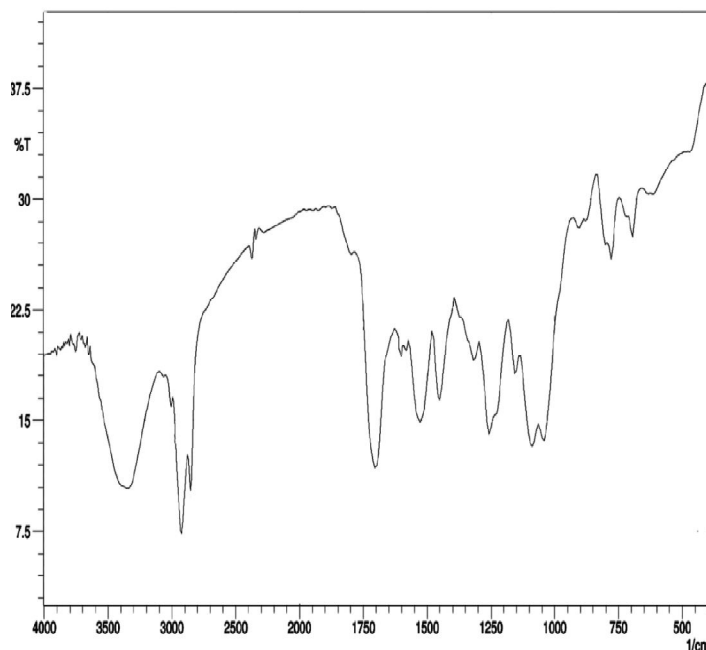


Figure 4.FT-IR spectrum of CDPGPU 10

### C. Thermal properties of polyurethane and its composites

The polyurethane and the corresponding composites (Figure 5 – Figure 7) decomposed around 90-95% in the temperature range of 450-500°C. This is due to the strong network formed between polyurethane prepared from 4,4'-methylenebis(cyclohexyl) isocyanate with cardanol based-polyol and glycol-glycoside.

The DTA thermogram does not show any endothermic peak for softening. But, two or three exotherms are invariably seen in both the urethanes. The first exotherm is relatively weak. However, the second and the third exotherms are strong. The first exotherm is attributed to the cleavage of long alkyl side chain of phenyl ring and also to the cleavage of allophanate linkages.

The TGA analysis of polyurethane composites based on cardanol based-polyol and polyol-glycoside possesses higher thermal stability when compared to neat polyurethane is due to higher crosslink density [26] and the data is presented in Table 2.

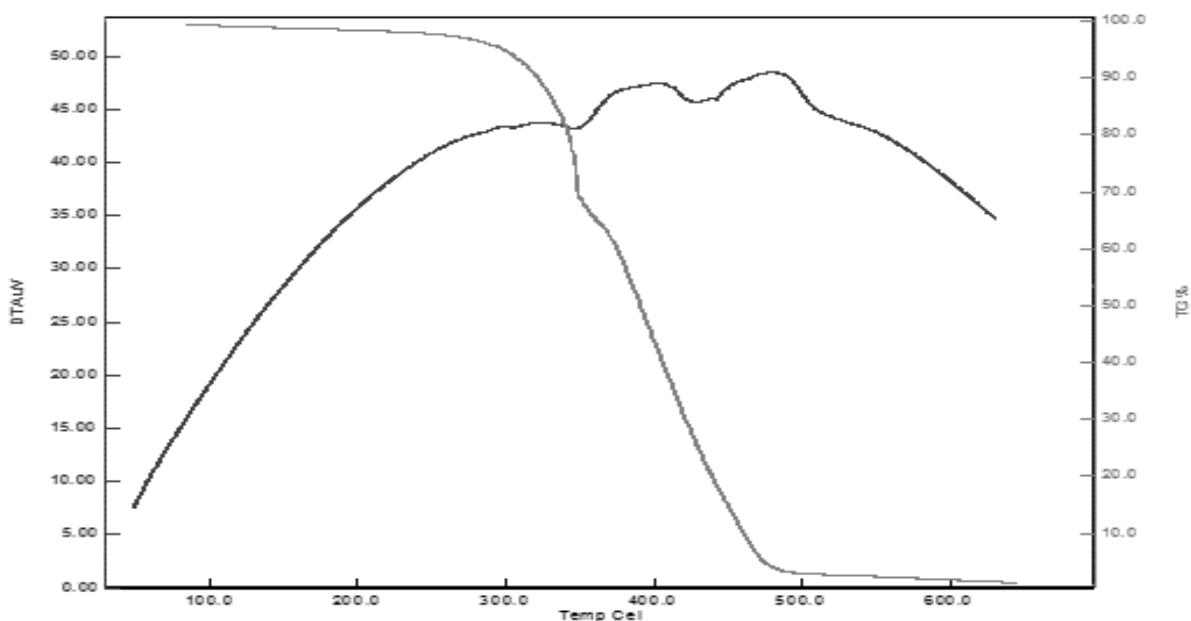


Figure 5.TGA/DTA of CDPGPU

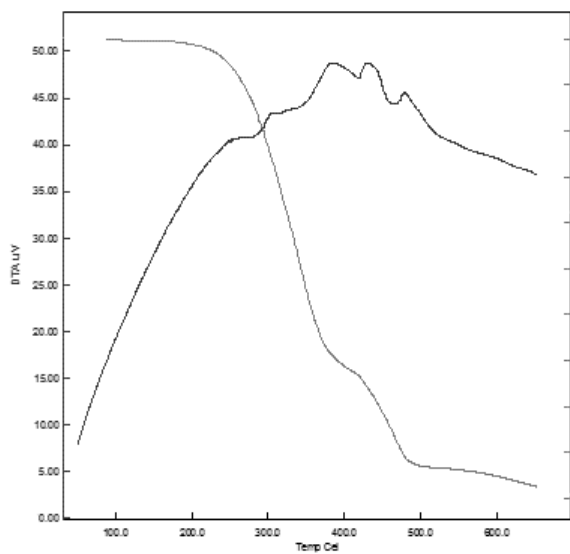


Figure 6.TGA/DTA of CDPGPU 5

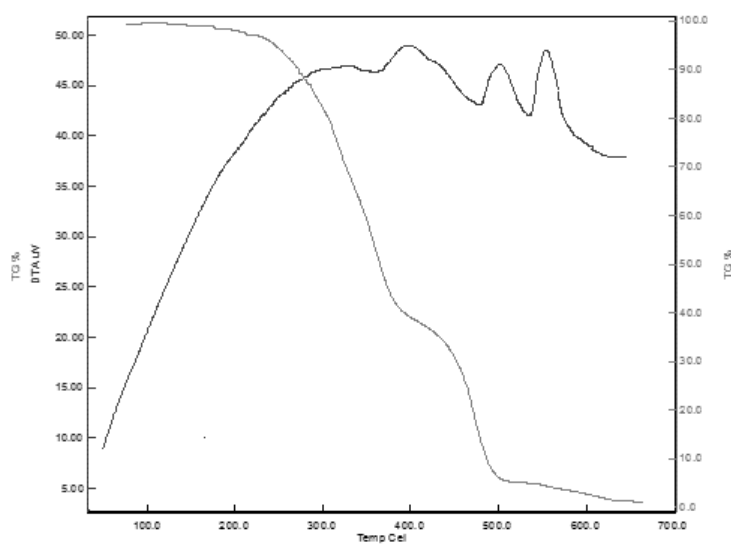


Figure 7.TGA/DTA of CDPGPU 10

Table 2.TGA/DTA data of polyurethane and its composites based on polyol-glycoside using coir fiber

Polyurethane and its composites	TGA data Temperature (°C) at the each stage of degradation (weight loss %)			DTA data		
	T <sub>start</sub> 1 <sup>st</sup>	T <sub>start</sub> 2 <sup>nd</sup>	T <sub>end</sub>	1 <sup>st</sup> exo	2 <sup>nd</sup> exo	3 <sup>rd</sup> exo
CDPGPU	292.4 (6.2)	341.5 (32.3)	5468 (94.5)	309.3	340.1	471.6
CDPGPU 5	234.5 (5.7)	362.6 (66.4)	481.2 (91.8)	306.3	378.4	476.2
CDPGPU 10	232.4 (4.7)	376.2 (63.2)	496.2 (92.1)	398	492.6	556.5

#### IV. CONCLUSIONS

Polyurethane composites prepared from cardanol based-polyol modified with polyol-glycoside and coir fiber showed excellent thermal stability when compared to the neat polyurethane. The glycoside based-polyurethane and the corresponding composites exhibited a uniform polyurethane phase without any presence of starch particles which reveals the presence of high hydroxyl content of polyol-glycoside. This is due to the higher level of degree of cross linking in these networks.

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