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Docking of Cathepsin D with New Compounds

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Abstract: Cathepsin D is a lysosomal protein belongs to aspartic protease family. Over expression of this protein is observed in certain pathological conditions like apoptosis, Alzheimer's disease, Cancer, Atherosclerosis, Metastasis of breast cancer etc. Some of these pathological conditions can be cured by inhibiting this protein. Present study was focused on screening and docking of several chemical compounds like Imidazoles, Oxadiazoles, Pyrazoles, Pepstatins, Piperazines with cathepsin d protein. Crystal structure of cathepsin D (1LYW) was extracted from PDB. These ligands were docked into the active site of Cathepsin D to study their binding energy and interactions. From these docking studies it was noticed that the pepstatin is the strong and very potent inhibitor of cathepsin d with estimated Inhibition Constant [Ki] 1.26 nM and -14.70 Kcal/mol Estimated Binding Energy. Next best inhibitor was Norchlorcyclizine with Ki 53.23 nM and -9.92 Kcal/mol Estimated Binding Energy. Remaining all other tested compounds were found to be having moderate binding capacity with high Ki in micro molars range. Key words: Cathepsin D, Docking, Cancer, Alzheimer disease, Metastasis.

I. INTRODUCTION

Cathepsin D mainly involves in the acidic degradation of lysosomal proteins. This protein is composed of two domains [6], which are separated by a large cleft. This cleft contains the active site that holds the substrate. Cathepsin D protein contains two catalytically active aspartic acid residues in its active site [Asp33 and Asp231] [7], which are contributed by each domain [6]. Cathepsin d active site can hold upto eight amino acids. Apart from these catalytically active aspartic acids, several amino acids from other regions in the protein forms hydrogen bonds with the substrate or inhibitors to strengthen the complex. By targeting Cathepsin d protein, we can study certain therapeutical areas like protein targeting [1], processing and presentation of antigens to MHC class II [2], muscular dystrophy [3], Alzheimer disease [4] degenerative brain diseases [5].

II. MATERIALS AND METHODS

- 1) Ligands and Protein Preparation: Crystal structure of cathepsin D (PDB: 1LYW) was collected from Protein Data Bank and the file was prepared by deleting all unwanted ligands and HETATM atoms. All the ligands structures were drawn using chemsketch tool. Both the ligands and protein energies were minimized and optimized. Ligands used for the study were belongs to the groups of Imidazoles, Oxadiazoles, Pyrazoles, Pepstatins, Piperazines.

III. RESULTS AND DISCUSSION

Cathepsin d protein active site was docked with Imidazoles, Oxadiazoles, Pyrazoles, Pepstatins, Piperazines. Other than the active site amino acids, few aminoacids were also strongly interacted with the ligands. Clotrimazole formed strong interaction with the aminoacids ARG 245, LYS 249, ALA 253, GLY 252. Norchlorcyclizine interacted with ASN 38, GLN 113, LEU 83, LEU 99, PHE 115, ILE 76, VAL 114, VAL 147, VAL 144. 3-(3-Bromophenyl)-1H-pyrazole interacted with TYR 86, ASN 70, HIS 45, SER 61. 1,2,4-Oxadiazole-3-methylamine interacted with ASP 62, LYS 63, ASP 62. Pepstatin interacted with LEU 83, VAL 144, VAL 147, TYR 78, SER 143, ILE 134, TYR 10, SER 36, ILE 142, ARG 202, SER 315. Binding Energy (Kcal/mol), Inhibition Constant Ki, hydrogen bond interactions were studied and the results were incorporated in the below tables.

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Table-1: Cathepsin d docking with Imidazoles

Calculatio	n of Binding Energy, Inhibitory co	onstant and of Imidazoles.	
	Estimated Binding Energy Kcal/mol	Estimated Inhibition Constant Ki.	H-bon
,	Kcai/illoi		H-00H

No Ligand Name nds -7.11 1. Clotrimazole $6.10\,\mathrm{uM}$ 5 2. Isoconazole -2.46 15.85 mM 6 3. Bifonazole -5.51 91.87 uM 4 4. Econazole -6.61 14.32 uM 4

The predicted binding energy of all the four Imidazoles were in between -2.46 Kcal/mol To -7.11 Kcal/mol and predicted inhibitory constant Ki values were in 6.10uM to 15.85 mM.

Table-II: Cathepsin d docking with Oxadiazoles Calculation of Binding Energy, Inhibitory constant of oxadiazoles

		Estimated Binding	Estimated	
No	Ligand Name	Energy Kcal/mol	Inhibition	H-bonds
			Constant Ki.	
1.	1,2,4-Oxadiazole-3-methylamine	-5.05	197.82 uM	6
2.	3-(chloromethyl)-5 methyl-1,2,4-oxadiazole	-3.19	4.62 mM	3
3.	3-Bromo-5-(bromomethyl)-1,2,4-oxadiazole	-3.59	2.32 mM	3
4.	5-phenyl-1,3,4-oxadiazole-2-thiol	-4.20	830.60 uM	6
5.	5-(4-Chlorophenyl)-1,3,4-oxadiazole-2-thiol	-4.56	457.05 uM	6
6.	5-methyl-1,3,4-oxadiazole-2-thiol	-3.10	5.32 mM	5

The predicted binding energy of all Oxadiazoles were in between -5.05 Kcal/mol to -3.10 Kcal/mol and predicted inhibitory constant Ki values were in between 197.82 uM to 5.32 mM.

Table-III: Cathepsin d docking with Piperazines: Calculation of Binding Energy, Inhibitory constant of Piperazines

		Estimated Binding	Estimated Inhibition Constant	
No	Ligand Name	Energy Kcal/mol	Ki.	H-bonds
1.	1-(2-Chlorophenyl) piperazine	-5.26	138.53 uM	6
2.	1-(2-Furoyl) piperazine	-4.15	900.48 uM	5
3.	Aminoethyl piperazine	-4.90	255.71 uM	9
4.	Norchlorcyclizine	-9.92	53.23 nM	6
5.	1-Isopropyl piperazine	-4.33	667.89 uM	4
6.	1-(4-pyridyl) piperazine	-4.54	470.72 uM	4

The predicted binding energy of all the Piperazines were in between -4.15 Kcal/mol to -9.92 Kcal/mol and predicted inhibitory constant Ki values were in between 53.23 nM to 900.48 uM.



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Table-IV: Cathepsin d docking with Pyrazoles Calculation of Binding Energy, Inhibitory constant of pyrazole

		Estimated Binding	Estimated Inhibition	
No	Ligand Name	Energy Kcal/mol	Constant Ki.	H-bonds
1.	1H-Pyrazole-3-carbaldehyde	-3.24	4.22 mM	7
2.	1H-pyrazole-4-carbaldehyde	-3.20	4.50 mM	7
3.	2,3-dihydro-1H-imidazo(1,2-b)pyrazole or	-3.33	3.60 mM	6
	Imidazolepyrazole			
4.	3-(3-Bromophenyl)-1H-pyrazole	-4.69	362.56 uM	6
5.	3-(4-Fluorophenyl)-5 (Methylthio)Pyrazole	-4.17	883.18 uM	5
6.	4-carboxypyrazole	-3.56	2.45 mM	7

The predicted binding energy of all the nine pyrazole were in between -3.20 Kcal/mol to -4.69 Kcal/mol and predicted inhibitory constant Ki values were in 362.56 uM to 4.50 mM.

Table-V: Cathepsin d docking with Pepstatins Calculation of Binding Energy, Inhibitory constant Pepstatins.

No	Ligand Name	Estimated Binding Energy Kcal/mol	Estimated Inhibition Constant Ki.	H-bonds
1.	Pepstatin	-14.70	1.26 nM	14
2.	Pepstatin analogue -1	-7.27	4.71 uM	12
3.	Pepstatin analogue -2	-8.19	1.00 uM	15
4.	Pepstatin analogue -3	-6.67	12.99 uM	9
5.	Pepstatin analogue -4	-5.20	153.45 uM	21
6.	Pepstatin analogue -5	-8.82	343.62 nM	4

The predicted binding energy of all the Pepstatins were in between - 5.20 Kcal/mol to -10.70 Kcal/mol and predicted inhibitory constant Ki values were in between 1.26 nM to 153.45 uM.

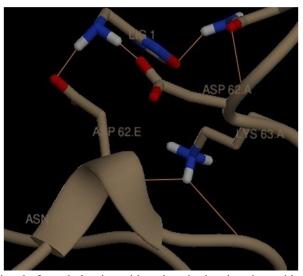


Fig: 1: 3D Interaction of 1,2,4-Oxadiazole-3-methylamine with cathepsin d active site residues, coloured straight lines representing the hydrogen bonds.

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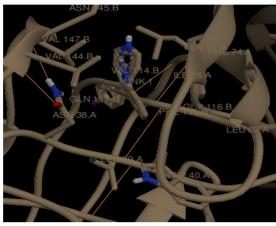


Fig:2 -3D Interaction of Norchlorcyclizine with cathepsin d active site residues, coloured straight lines representing the hydrogen bonds formation.

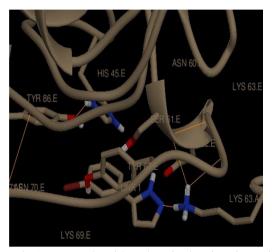


Fig: 3- 3D Interaction of 3-(3-Bromophenyl)-1H-pyrazole with cathepsin d active site residues, coloured straight lines representing the hydrogen bonds formation.

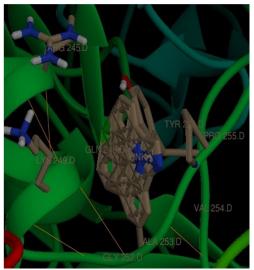


Fig: 4 - 3D Interaction of clotrimazole with cathepsin d active site residues, coloured straight lines representing the hydrogen bonds formation

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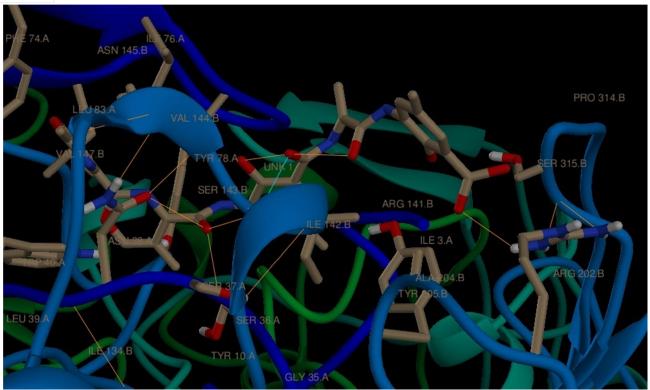


Fig: 5 - 3D Interaction of Pepstatin with cathepsin d active site residues, coloured straight lines representing the hydrogen bonds formation

IV. CONCLUSION

Apart from Pepstatin, Norchlorcyclizine was observed to be inhibiting cathepsin d in nanomolar range, and few amino acids in cathepsin d are actively interacting with inhibitor molecules are VAL 114, VAL 147, VAL 144, SER 143, ILE 134, TYR 10, SER 36. These docking studies helps in the exploration of cathepsin d active site aminoacids which takes part in the binding and interaction with the inhibitor molecules. By studying wide range of compounds as cathepsin d inhibitors, helps us in the development of new therapeutic agents by focusing on new synthetic approaches.

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